# organic papers

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#### Key indicators

Single-crystal X-ray study T = 298 K Mean  $\sigma$ (C–C) = 0.010 Å R factor = 0.081 wR factor = 0.168 Data-to-parameter ratio = 8.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 3,4,6-Tris(pyrazol-1-yl)pyridazine

In the structure of the title compound,  $C_{13}H_{10}N_8$ , a tetradentate N<sub>4</sub>-donor ligand derivatized on the pyridazine backbone with a monodentate N-donor group, the four potentially coordinating N atoms of the pyridazine and 3-and 6-pyrazole rings adopt a *trans-trans* conformation. Although the 6-substituted pyrazole ring is almost coplanar with the pyridazine ring, the 3- and 4-substituted pyrazole rings are severely bent out of the plane of the pyridazine ring. These features suggest that it may not be possible to arrange the four adjacent N-donors such that the molecule can act as a bis-bidentate chelating ligand. An analysis of the extended structure of the title compound reveals a very short, offset face-to-face  $\pi$ - $\pi$  interaction involving the pyridazine and 6-substituted pyrazole rings of adjacent molecules.

## Comment

Pyridazines substituted in the 3- and 6-positions with N-donor ligands act as tetradentate N<sub>4</sub>-donor ligands in a bis-bidentate chelating fashion, generating multinuclear coordination complexes with relatively short internuclear separations  $[d(M \cdots M) \ ca \ 3.6 \ \text{\AA};$  Thompson *et al.*, 1985; Youinou *et al.*, 1992, Hubberstey & Russell, 1995].



In an attempt to introduce a third ligating centre to these molecules, we have prepared 3,4,6-tris(pyrazol-1-yl)pyridazine, (I), a bis-bidentate N<sub>4</sub>-donor ligand derivatized on the pyridazine backbone with a monodentate N-donor group. Its molecular structure is shown in Fig. 1. Two noteworthy points emerge. Firstly, the four potentially coordinating N atoms of the pyridazine and 3- and 6-pyrazole rings (N1, N2, N32 and N62) adopt a trans-trans conformation, which contrasts with the cis-cis conformation required for the tetradentate N<sub>4</sub>-donor ligands to act in a bis-bidentate chelating fashion. Secondly, although the 6-substituted pyrazole ring is almost coplanar with the pyridazine ring [dihedral angle 5.5  $(4)^{\circ}$ ], the 3- and 4-substituted pyrazole rings are severely bent out of the plane of the pyridazine ring [dihedral angles 40.2 (3) and 51.2 (2)°, respectively]. Unfortunately, both points, but especially the latter, which can be attributed to steric conflict between the adjacent pyrazole substituents on the 3- and 4-positions of the pyridazine ring, suggest that it Received 29 October 2002 Accepted 14 November 2002 Online 22 November 2002

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Figure 1

The molecular structure and atom-numbering scheme of 3,4,6-tris(pyrazol-1-yl)pyridazine. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as spheres of arbitrary radii.





A projection of the structure of two molecules onto the least-squares mean plane containing the pyridazine and 6-substituted pyrazole rings of one molecule, showing the  $\pi$ - $\pi$ -stacking interactions between adjacent molecules.

may not be possible to arrange the four adjacent N-donor centres for it to act as a bis-bidentate chelating ligand.

An analysis of the extended structure reveals the existence of a very short, offset face-to-face  $\pi$ - $\pi$  interaction involving the pyridazine and 6-substituted pyrazole ring of adjacent molecules (Fig. 2). The perpendicular separation between the least-squares mean planes through these two rings [the maximum deviation of fitted atoms from best plane is 0.10 Å] is very short [3.339 (12) Å; range 3.244-3.409 Å]. As each tris(pyrazol-1-yl)pyridazine molecule forms part of a weakly C-H···N hydrogen-bonded chain [C34-H34 = 0.93, H34···N42 = 2.62, C34···N42 = 3.515 (8) Å and C34-H34···N42 = 162°; C35-H35 = 0.93, H35···N62 = 2.50, C35···N62 = 3.413 (8) Å and C35-H36···N62 = 168°] aligned in the [101] direction (Fig. 3), the offset  $\pi$ - $\pi$  interactions link the chains to give a three-dimensional matrix (Fig. 4).

## **Experimental**

Sodium hydride (0.98 g, 245 mmol) was added to a solution of pyrazole (1.12 g, 165 mmol) in pre-dried tetrahydrofuran (50 ml). After stirring the mixture for 20 min, 3,4,6-trichloropyridazine





A projection of the structure on to the (010) plane, showing the C34– $H34\cdots N62$  and C35– $H35\cdots N42$  hydrogen-bonding interactions, which generate the chain of molecules aligned along the [101] direction. Key: C black circles, N blue circles and H small yellow circles.





A view of the structure showing the  $\pi$ - $\pi$ -stacking interactions linking the hydrogen-bonded chains. Key: C black circles, N blue circles and H small yellow circles.

(1.00 g, 54 mmol) was added to the solution (CAUTION: exothermic reaction!) and the mixture stirred for a further 60 min. After cooling to room temperature, the solvent was removed and the resultant solid dissolved in dichloromethane (40 ml) and washed with water (3  $\times$  30 ml). The organic layer was dried over magnesium sulfate and the solvent removed to give a white powder, which was recrystallized from ethanol (yield; 1.32 g, 47 mmol, 88%) to give crystals suitable for diffraction analysis. Found (calculated for C<sub>13</sub>H<sub>10</sub>N<sub>8</sub>): C 55.90 (56.10), H 3.60 (3.60), N 40.20% (40.25%). IR (KBr disc) ( $\nu/cm^{-1}$ ):

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3132 (*m*), 1594 (*s*), 1562 (*s*), 1526 (*s*), 1456 (*s*), 1424 (*s*), 1396 (*s*), 1336 (*s*), 1323 (*m*), 1198 (*s*), 1189 (*m*), 1175 (*s*), 1113 (*m*), 1098 (*m*), 1059 (*s*), 1045 (*s*), 1031 (*s*), 1015 (*s*), 954 (*s*), 939 (*s*), 902 (*s*), 893 (*s*), 865 (*m*), 811 (*s*), 777 (*s*), 759 (*s*), 669 (*m*), 649 (*m*), 624 (*s*), 599 (*s*), 584 (*m*), 521 (*m*), 487 (*m*), 443 (*m*). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ /p.p.m.: 6.45 (*m*, 1H), 6.68 (*m*, 2H), 6.80 (*d*, 1H), 7.81 (*d*, 1H), 7.84 (*d*, 1H), 7.93 (*d*, 1H), 8.21 (*dd*, 1H), 8.71 (*s*, 1H), 8.84 (*dd*, 1H). EI–MS (*m*/*z*) 278 [C<sub>13</sub>H<sub>10</sub>N<sub>8</sub>]<sup>+</sup>.

Crystal data

 $\begin{array}{l} C_{13}H_{10}N_8\\ M_r=278.29\\ Monoclinic, P2_1/n\\ a=13.190~(5)~\AA\\ b=7.003~(3)~\AA\\ c=14.326~(4)~\AA\\ \beta=102.14~(3)^\circ\\ V=1293.7~(8)~\AA^3\\ Z=4 \end{array}$ 

#### Data collection

Stoe Stadi-4 four-circle diffractometer  $\omega/\theta$  scans Absorption correction: none 3374 measured reflections 1688 independent reflections 856 reflections with  $I > 2\sigma(I)$  $R_{int} = 0.117$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.081$   $wR(F^2) = 0.168$  S = 1.191688 reflections 190 parameters H-atom parameters constrained  $D_x = 1.429 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 29 reflections  $\theta = 10.0-12.0^{\circ}$  $\mu = 0.10 \text{ mm}^{-1}$ T = 298 (2) KPlate, colourless  $0.23 \times 0.23 \times 0.02 \text{ mm}$ 

 $\begin{array}{l} \theta_{\max} = 22.5^{\circ} \\ h = -14 \rightarrow 14 \\ k = 0 \rightarrow 7 \\ l = -15 \rightarrow 15 \\ 3 \text{ standard reflections} \\ \text{frequency: 60 min} \\ \text{intensity decay: none} \end{array}$ 

$w = 1/[\sigma^2(F_o^2) + (0.014P)^2]$
+ 1.704P]
where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} = 0.001$
$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.27 \text{ e} \text{ Å}^{-3}$

This crystal diffracted only to low resolution. No significant diffraction occurred beyond  $2\theta$  of  $45^{\circ}$ , which accounts for the high value of  $R_{\text{int}}$  (0.117). All H atoms were included at geometrically calculated positions and constrained to ride at a distance of 0.93 Å from their parent C atoms, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *STADI*4 (Stoe & Cie, 1997); cell refinement: *STADI*4; data reduction: *X-RED* (Stoe & Cie, 1997); program(s) used to solve structure: *SIR*92 (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *SHELXL*97 and *PLATON* (Spek, 2002).

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